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IMA- Question Bank for Test - 1

Portions:

Unit-I

- Beer-Lambert's law, limitations, derivation of the law from first principles
- Applications of Beer-Lambert's law to single component and multicomponent measurements
- Principle and instrumentation of UV-vis spectroscopy
- Estimation of inorganic ions such as Fe, Ni and Nitrite using Beer-Lambert's Law
- Principle and instrumentation of Infrared spectroscopy
- Electromagnetic radiation regions and properties, interaction of photons with matter
- Woodward-Fischer rules for the calculation of absorption maxima

Unit-V

- Chromatographic methods principles, types, improving the resolution of separation
- HPLC
- GC
- Estimation of organic compounds by HPLC and GC

Part A (2 mark questions)

- 1. Give the relationship between wavelength and energy of electromagnetic radiation.
- 2. Convert the following:
 - (a) 20% transmittance into absorbance scale.
 - (b) 1.4 absorbance into % transmittance scale.
- 3. Convert the wave number 3500 cm^{-1} to wavelength in $^{\circ}A$.
- 4. Discuss few applications of UV and Visible spectrophotometry.
- 5. Calculate the absorptivity of a compound with molecular weight 144, if 10^{-5} (g/ml) of solution exhibits an absorbance of 0.4, when the optical path is 1 cm.
- 6. Give the spectral range of of UV-vis spectrometers.
- 7. What are the main parts of a spectrometer?
- 8. What are the light sources used in UV-vis spectroscopy?
- 9. What are the radiation sources involved IR spectroscopy?

10.

- 11. What standard is used to check the correctness of IR instrument?
- 12. What standard is used to check the correctness of HPLC instrument?
- 13. Enlist the typical carrier gases used in GC.
- 14. Differentiate between isocratic and gradient elution in HPLC.
- Dr. M. Subramanian, Associate Professor, Chemical Engineering, SSN College of Engineering, Chennai.



- 15. What are the detectors used in HPLC?
- 16. What are the types of columns used in GC?
- 17. What are the detectors used in GC?
- 18. How is the presence of a component identified in GC?
- 19. How is the concentration of a sample quantified with HPLC?
- 20. What are the methods by which the resolution of peaks in GC and HPLC can be improved?
- 21. Mention the typical solvents used in HPLC.

Part B (16 marks questions)

Explain with a schematic, the principle, applications, instrumentation and typical output of: (i) UV-vis spectroscopy, (ii) IR spectroscopy, (iii) HPLC, (iv) GC

